

Analysis of Corrosion Behaviour In Industries

Pooja Kushwaha¹, Swati Soni²

^{1,2}Dept of CSE

^{1,2} Takshshila Institute of Engineering & Technology, Jabalpur, MP, India

Abstract- Corrosion is an electro chemical reaction that occurs if the coating gets damaged even in a small area or if it develops cracks or if the thickness of the coating is not adequate. The metal pipeline, thus exposed to soil or water, will form two electrodes of opposing polarity (anodes and cathodes) on the same metal. This may be due to chemical composition, velocity, temperature, stress or impurities in the metal. The electrical potential difference between these newly formed anodes and cathodes gives rise to a current that causes the pipe line to lose metal and become leaky. The corrosion develops and begins to spread, causing enormous damage to property as well as environment. The loss due to Corrosion has been estimated to be of the order of 2 - 5% of the Gross National Product (GNP) of any Country.

Keywords- SCADA, PLC, X-Ray Diffraction.

I. INTRODUCTION

Modern procedures will be methods including compound, physical, electrical or mechanical strides to help in the assembling of a thing or things, for the most part did on an exceptionally enormous scale. Modern procedures are the key segments of overwhelming industry. Assembling is the generation of items for use or deal utilizing work and machines, devices, substance and organic handling, or plan. The term may allude to a scope of human action, from handiwork to cutting edge, yet is most ordinarily applied to modern structure, in which crude materials are changed into completed products on a huge scale. Such completed merchandise might be offered to different makers for the generation of other, increasingly complex items, for example, flying machine, family unit apparatuses, furniture, athletic gear or cars, or offered to wholesalers, who thusly offer them to retailers, who at that point offer them to end clients and customers. Assembling building or assembling procedure are the means through which crude materials are changed into a last item. The assembling procedure starts with the item plan, and materials detail from which the item is made. These materials are then adjusted through assembling procedures to turn into the necessary part. Current assembling incorporates every middle of the road procedure required in the generation and combination of an item's segments ^[9].



Fig-1 Industrial Processes and Manufacturing

II. BACKGROUND

Carbon steel and austenitic stainless steels are the most widely used construction materials to handle sulphuric acid solutions. Carbon steel is an inexpensive alloy; however, its usage requires a comprehensive understanding of its corrosion behavior in corrosive environments ^[6]. Rubber coatings are often used to protect carbon steel equipment against corrosion in hydrometallurgical circuits. A major failure of the rubber coatings is the disbonding of the coating due to chemical and mechanical actions. The disbondment of the coating will form a gap between the coating and the carbon steel substrate that may significantly increase the tendency of the substrate for degradation via general, pitting and crevice corrosion. Pitting and crevice corrosion are considered important concerns if metals are exposed to chloride containing solutions. Several factors affect the occurrence and rate of crevice and pitting corrosion including alloy composition, geometry, pH, dissolved ions concentrations and temperature ^[1].

Importance of Corrosion

Corrosion is a crucial factor that limits the lifetime of many metallic structures and it's one among the first means by which materials deteriorate, especially in humid and saline environment ^[8]. Corrosion is an accumulative process that alters the surface property and fracture strength of structures, leading to premature deterioration. Because the structural components ages, the occurrence of corrosions increases.

Additionally, corrosion occurrence is independent of usage. Inspection for corrosions is time consuming, costly and sometimes causes additional damages since corrosions mostly occur at inaccessible locations that can't be inspected visually. Corrosion may be a persistent problem which will only be managed but not eliminated. The impacts on cost saving and safety assurance would be significant if hidden corrosions are often detected without disassembly [8].



Fig-2 Loss from Corrosion

The X-ray diffraction analysis showed that the corrosion products layer mainly consisted of lepidocrocite (γ -FeOOH), goethite (α -FeOOH) and iron sulphide (FeS) on the crevice edges. The Pourbaix graph of iron in sulphuric corrosive arrangement at temperature demonstrates that iron sulfide is formed on the metal surface at various pH esteems. The akaganeite (β -FeOOH) diffraction pinnacle wasn't distinguished in any range which might be due to the low centralization of chloride particles within the arrangements. Besides, the number of lepidocrocite pinnacles diminished because the sulphuric corrosive fixation expanded from 10g to 50g. The lepidocrocite is choppy within the sight of sulphuric corrosive, and therefore the disintegrated particle goes about as an oxidant to the metal and henceforth lower lepidocrocite pinnacles are distinguished. Electrochemical commotion estimation (ECN) testing was likewise performed to look at the erosion procedure happening on the steel surface with the designed cleft. The aftereffects of the ECN estimations demonstrated that current expanded during initial few minutes and then diminished marginally. Additionally, the coupled potential didn't change after an underlying movement negative way. The low current moving through the steel cathodes and therefore the steady potential demonstrated that the opening erosion didn't create. These outcomes infer that the fissure consumption might not happen on the steel surface in acidic arrangements containing chloride particles [4].



Fig-2 Creature Collection & Pattern Creation

The crevice corrosion is difficult to detect since it remains a dangerous phenomenon under the disbonded coating. It's a sort of localized attack that happens in areas of the metal or alloy surface in touch with a stagnant solution while the foremost a part of the surface is exposed to the majority solution. Although the crevice corrosion may occur on the metal surfaces which aren't covered by passive films, but it occurs frequently on passivated metal surfaces. However, pitting corrosion occurs only at the passivated metal surfaces [13]. It suffering from concentration of oxygen and hydrogen ions, humid and dry environment conditions, electrochemical potential, dissolved ions in solution, and thermodynamic stability of the oxides and oxy-hydroxides.

III. CHARACTERISTICS OF CORROSION PRODUCTS ON STEEL SURFACE

Corrosion products formed on a steel surface mainly contains various iron oxides, iron hydroxides, and iron oxy-hydroxides. The morphology and phase composition of the corrosion products are important since the corrosion rate and properties of substrates are strongly suffering from the characterization of the corrosion products layer. The corrosion process of iron in aqueous solutions includes re-oxidation and reduction of iron (II) and iron (III) ions. The iron is dissolved as iron (II) ions into the answer, and these ions are precipitated onto the metal surface because the corrosion products layer [14]. Moreover, a high concentration of chloride ions causes the formation of akaganeite (β -FeOOH) with an acicular shape structure.



Fig-3 Corrosion Products on steel Surface

IV. CREVICE CORROSION BEHAVIOUR OF STEEL

Crevice corrosion may be a localized attack in acid-chloride conditions occurring in prone regions of metals and alloys like under deposits, beneath gaskets, in imperfect welding, interface of washers, and within pits and cracks. Crevice corrosion affects metals and alloys that form a protective thin oxide layers on their surface. A little current called the passive current, can undergo the protective film since the film doesn't have a high electrical conductivity. Therefore, within the presence of the protective film, metal dissolution occurs at a really slow rate. However, in concentrated acid-chloride solutions, the protective film is destroyed when the answer pH and chloride concentration reach a critical state that's called the critical crevice solution [15].



Fig-4 Schematic diagram of crevice corrosion

V. PITTING CORROSION BEHAVIOUR OF STEEL

Pitting corrosion happens at the metal surface covered with the passive or protective layer (a thin oxide layer) during the access of aggressive anions, especially chloride ions. A critical potential called the pitting potential must be exceeded for the initiation of pitting corrosion [2].



Fig-5 Pitting Corrosion Behaviour of steel

VI. COMPARISON AND ANALYSIS

Pradeep Jain, Umesh C. Bhakta and Salil K. Sanyal et al [2, 2000], presents a contextual investigation led to get the explanations of break/disappointment at twists of water cooled stator bar during a 210 MW generator. The bars containing empty and powerful sub-conductors were cleaned

to expel the protection. The bombed segments were inspected under examining electron magnifying instrument, optical magnifying lens and therefore the erosion items by X-beam examination. The examination uncovered the nearness of buries granular weak splits, cuprous oxide stain film, formulation, and dampness. of these demonstrated the trademark highlights of SCC in copper metal in ammonical arrangement. A definitive burst by SCC of strong and empty conductors occurred in three phases:

Sascha Schmeling et al [3, 2006], the Joint Controls Project (JCOP) may be a communitarian exertion by the four LHC tests and a couple of care groups within the examination division of CERN. It had been initially started by a suggestion from the Working Group on Common Projects within the Field of Software/Computing and Trigger/DAQ at the massive Hadron Collider (LHC). The command of this undertaking was characterized in an archive in 2000, yet has been reached out during the foremost recent years, incompletely due to an outdoor survey in 2003. One among the many increases has been the incorporation of the assistance and maintenance plan.

Pavel A. Lyakhov, Anzor R. Ozaev, Nikolay I. Chervyakov and Dmitrii I. Kaplun et al [10, 2019], new method of adaptive median filtering of impulse noise in images is proposed during this paper. The method is predicated on the combined using of iterative image processing and post-processing of the median filtering result. The results of a comparison of the standard of the proposed method with known methods are shown within the experimental a part of the article. The simulation results showed that the proposed method performs better than known methods altogether cases. The results obtained within the paper can find wide application within the processing of satellite and medical images, geophysical data and other applications of digital image processing.

VII. CONCLUSIONS

Several corrosion measurement tests are employed to review the corrosion processes and corrosion characteristics of varied metals and alloys in aqueous environments. They supply important information on the corrosion rate, corrosion mechanism, passivity and coatings by measuring the potential-current relations under certain conditions. This section explains the theories of the corrosion techniques, which were employed during this scientific research. Pradeep Jain, Umesh C. Bhakta and Salil K. Sanyal et al [2], presents a contextual investigation directed to discover the reasons of crack/disappointment at curves of water cooled stator bar in a 210 MW generator. The bars containing empty and strong sub-conductors were cleaned to expel the protection. The

bombed parts were inspected under filtering electron magnifying lens, optical magnifying instrument and the consumption items by X-Ray investigation. This disadvantage of the method is this; it requires a number of equipment even it consumes more time.

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