# Performance Study of Yttria Stabilized Zirconia And Gadolinium Zirconate Coating For Nickel Base Alloy For Turbine Application

**Dr. G. Purushotham** 

Professor & HOD, Dept. of Aeronautical Engg Mangalore Institute of Technology & Engineering, Moodabidri, Mangalore, Karnataka, India. Pin Code: 574225

Abstract- Experimental work is carried out to deals with the development of new synthesis techniques for the functional materials such as Yttria Stabilized Zirconia (YSZ) and Gadolinium Zirconate used in the field of thermal barriers coatings. Currently, Thermal Barrier Coatings (TBCs) are manufactured by dry route technologies, but such methods are directional and often require costly investments and complicated operations. We have carried out significant work aimed at developing sol-gel routes, which are nondirectional methods, to prepare, by suitable chemical modifications, nanocrystalline materials with a controlled morphology. The main advantage of this method is to decrease the crystallization temperature, much lower than the conventional processes, allowing the synthesis of reactive powders with nanometric particles size. In this research work, the formulation of an alkoxide sol has been optimized in order to obtain homogeneous YSZ films. The films microstructures have been investigated using scanning electron microscopy and thermal property have been studied.

*Keywords*- Nickel alloy, Yttria Stabilized Zirconia, Gadolinium Zirconate, thermal barriers coatings.

# I. INTRODUCTION

Thermal barrier coating systems (TBCs) are widely used in modern gas turbine engines to lower the metal surface temperature in combustor and turbine section hardware. Engines for both aero-jet propulsion and land based industrial power generation have taken advantage of this technology to meet increasing demands for greater fuel efficiency, lower Noxemissions, and higher power and thrust [1-2]. The engine components exposed to the most extreme temperatures are the combustor and the initial rotor blades and nozzle guide vanes of the high-pressure turbine [3]. Metal temperature reductions of up to 165<sup>o</sup>C are possible when TBCs are used in conjunction with external film cooling and internal component air cooling. In film cooling a protective blanket of cooling air is ejected onto the external surface of the turbine vane or blade, from internal passages within the airfoils, by means of holes or slots in the surface. A typical TBC consists of two key layers: an oxidation resistant bond coat such as diffusion aluminide or overlay MCrAlY bond coating, and a ceramic top layer, typically 7-8 wt.% Y23-stabilized ZrO2(7YSZ), to reduce the heat flux into the component [4-5].

The ceramic top layer is typically applied either by plasma spray or by electron beam physical vapor deposition (EBPVD) [6]. Significant effort is going into the development of new ceramic compositions with lower thermal conductivity. Concepts used are advanced multicomponent zirconia (ZrO2)based TBCs using an oxide defect clustering design and materials whose compositions have a pyrochlore structure. In addition to a low-thermal conductivity, other desired properties of a thermal barrier coating are also phase stability during long term high-temperature exposure and thermal cycling, resistance to erosion and pollutants such as, e.g., CMAS, which can destroy the integrity of the thermal barrier coating by infiltrating the pores and react with the coating [7].

#### **II. EXPERIMENTAL WORK**

# Synthesis of Yttria Stabilized Zirconia and Gadolinium Zirconate:

Synthesis of YSZ and  $Gd_2Zr_2O_7$  requires Mortar and Pestle .For 7 wt % of YSZ component  $ZrO_2$ :  $Y_2O_3$  should be in 0.93 :0.07 Molar ratio, which are added to the mortar for uniform mixing .Pestle helps in crushing the components against the mortar into further nano particles .To ensure proper mixing acetone/alcohol is added in a dropwise at an interval of 15 minutes .This process is continued for duration 4-5 hours continuously .At the end of process, the component is available in the form of finely divided powder which is further kept for calcination at 1000° C for 2 hours .The same process is performed for synthesis of Gadolinium Zirconate but the calcination temperature is changed to 1500° C.

#### ISSN [ONLINE]: 2395-1052



Fig 1: Coating Powder developed at laboratory

The microstructure morphology of coating material is investigated by a field emission scanning electron microscope )SEM (and the phase analysis is carried out by X-ray Diffraction )XRD (with filtered Cu K $\alpha$  radiation using laboratory facilities available at NITK Surathkal .



Fig 2: JOEL Scanning Electron Microscope

A scanning electron microscope )SEM (uses electrons as a probe just like an optical microscope uses light as a probe .The investigation is utilized in imaging and chemical analysis and forms the subject of scanning electron microscopy .The SEM chamber operates at high vacuum ) $10^{-4}$ -100Pa( the specimens are mounted on stub with 2-sided carbon tape .Adjust the stage positioning using XY knobs and note specimen position and height and place the specimen gently push the stage door back .Adjust the parameters to get best image and valid spectroscopy.



Fig 3: JEOL X-ray Diffractometer

X-ray powder diffraction )XRD (is a rapid analytical technique primarily used for phase identification of a crystalline material and can provide information on unit cell dimensions .The analyzed material is finely ground, homogenized, and average bulk composition is determined .X-ray diffractometers consist of three basic elements :An X-ray tube, a sample holder, and an X-ray detector .

X-rays are generated in a cathode ray tube by heating a filament to produce electrons, accelerating the electrons toward a target by applying a voltage, and bombarding the target material with electrons .When electrons have sufficient energy to dislodge inner shell electrons of the target material, characteristic X-ray spectra are produced.

# **III. COATING PROCESS**

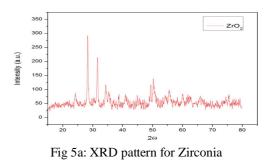
Before the coating process begins the substrate has to be cleaned and obtain dust free .So, the substrate was grounded with SiC paper upto  $100\mu m$ .They were then cleaned with a series of acetone, methanol and distill water in an ultrasonic set .In order to remove moisture and other adsorbents and improve coating adhesion, the substrate was preheated at 95°C for 5 minutes.



Fig 4: Ultrasonic setup to clean the substrate

The necessary coating paste was made by addition of polyvinyl alcohol to the coating powder YSZ in a mortar to the mixture an appropriate amount of absolute alcohol is added .Using the paint brush the coating paste produced is applied on to the surface of substrate material and left undisturbed for an hour to get dry .In order to complete the coating process, the coated substrate is kept in a furnace at 1000°C and the samples were held at this temperature for 5 hours. After the temperature of samples were let cool to room temperature and later testing and analysis on specimens are carried out.

#### **IV. RESULT AND DISCUSSION**



www.ijsart.com

# IJSART - Volume 3 Issue 9 -SEPTEMBER 2017

XRD pattern is taken for ZrO2 for phase analysis with filtered Cu Ka radiation .The obtained XRD pattern is compared with a standard pattern of similar ZrO2 and by referring the characteristic peaks, hence we concluded that  $ZrO_2$  exhibit tetragonal phase further it can be verified through Bragg's Law satisfying the formula  $n\lambda = 2d$ )sin $\theta$ .(

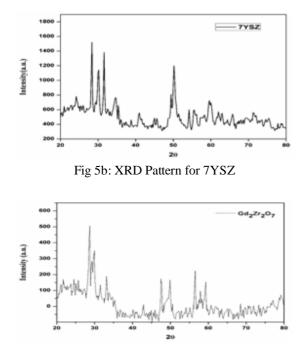


Fig 5c: XRD image of Gadolinium Zirconate

From the above shown figures obtained by X-ray diffraction calcined at 1000°C shows for 7YSZ and 8YSZ phase transformation from cubic to tetragonal .The characteristic peaks obtained are correlated with the reported XRD pattern shows similar representation confirming the above said. The XRD pattern of  $Gd_2Zr_2O_7$  shows a pyrochlore and the additional peaks incorporated are the indications of impurities present in specimens.

# MICROSTRUCTURE ANALYSIS BY SEM

The morphology of three feedstock shown through SEM images it is that 7YSZ and 8YSZ prepared by solid state reaction depicts well grown spherical grains with dense microstructure with varying size 80-100 nm .Whereas Gd<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> prepared by co-precipitation technique represents the small nano grains of approx .96-115 nm with dense microstructure.

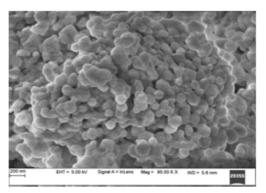


Fig 6a: SEM image of 7YSZ

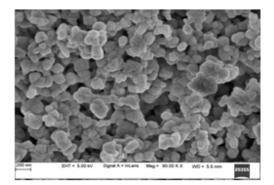


Fig 6b: SEM image of 8YSZ

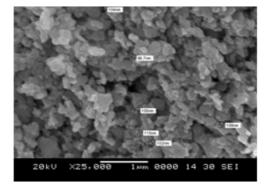


Fig 6c: SEM image of Gd<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub>

# THERMAL CONDUCTIVITY

Thermal Conductivity measured for different specimens were recorded in tabular column and graph is plotted for the same.

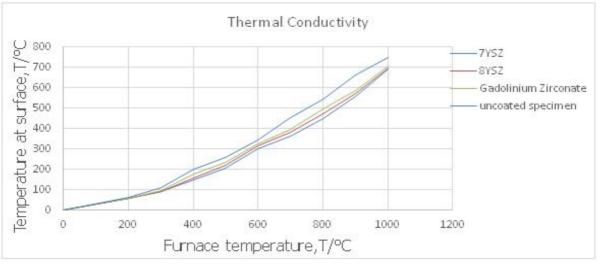


Fig 7: Thermal conductivity comparison of various specimens

The graph obtained from the data recorded, performed through measuring the surface temperature of the coated specimen at corresponding temperature of the furnace shows that 7YSZ has lower thermal conductivity compare to 8YSZ and Gadolinium Zirconate .While Uncoated specimen found to have higher thermal conductivity value than coated specimens.

#### **V. CONCLUSIONS**

During the experimental work we developed a new synthesis technique for functional material such as Yttria Stabilised Zirconia )YSZ (and Gadolinium Zirconate used in the field of thermal barrier coatings .We have carried out significant work aimed at developing sol gel routes, which are nondirectional methods to prepare by suitable chemical modifications, nanocrystalline material with controlled morphology .The films microstructure investigated using scanning electron microscopy, the images of particles seem to be dense with small nano grain size ranging from 90-100nm. The X-ray diffraction data collected are referred with reported data available through literatures indicates that characteristic peaks in a XRD pattern of 7YSZ and 8YSZ exhibit tetragonal phase while Gd<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> exist as pyrochlore and other interrupted peaks shows the presence of impurities among the samples .The conductivity of 7YSZ is lower compared with 8YSZ, Gd<sub>2</sub>Zr<sub>2</sub>O<sub>7</sub> and substrate uncoated .This might be due to a lower density for a pallet prepared for 7YSZ compared to other pallets.

## REFERENCES

 Miller R A (2007) Thermal barrier coatings for aircraft engines :history and directions *Journal of Thermal Spray Technology*, Vol. 139, pp 32–39.

- [2] L.Li, D.R. Clarke, (2009) Effect of CMAS Infiltration on Radiative Transport Through an EB-PVD Thermal Barrier Coating, Inter *-national Journal of Applied Ceramic Technology*, Vol. 74, pp 111–113.
- [3] J.Wu, H.B.Guo, Y.Z.Gao, et al.,(2008) Microstructure and thermo -physical properties of yttria stabilized zirconia coatings with CMAS deposits, *Journal of the European Ceramic Society*, Vol. 75, pp 86–89.
- [4] S.H. Song, P. Xiao (2013) An impedance spectroscopy study of high -temperature oxidation of thermal barrier coatings, *Materials Science and Engineering*, Vol. 67, pp 79-91.
- [5] J. Wu, H.B. Guo, L. Zhou, L. Wang, et al., (2014) Microstructure and thermal properties of plasma sprayed nanostructure YSZ thermal barrier coatings, *Journal of Thermal Spray Technology*, Vol. 39, pp 174–181
- [6] B.Jayaraj, V.H.Desai, C.K.Lee, et al.,) 2000 ( Electrochemical impedance spectroscopy of porous ZrO<sub>2</sub>-8 wt %.Y<sub>2</sub>O<sub>3</sub> and thermally grown oxide on nickel aluminide, *Materials Science and Engineering*, Vol. 83, pp 807–811
- [7] Vassen R, Traeger F, Stöver D )2011 (New Thermal Barrier Coatings Based on Pyrochlore/YSZ Double Layer Systems *International Journal of Applied Ceramic Technology*, Volume 11, <u>Issue 3</u>, pp 349–358.